

NAME:

Unit 4 In-class Isotope Practice

Symbol	Protons	Neutrons	Electrons	Charge
	27	31		0
		67	50	0
$^{209}_{82}\text{Pb}$				
		11	10	+1
	39	51		0
		42	30	+3
$^{240}_{93}\text{Np}$				
$^{25}_{12}\text{Mg}$				
$^{34}\text{S}^{-2}$				
	104	158		0
	10	20	10	
	7	16		-3
$^{28}_{13}\text{Al}$				
		76	54	-1
^{199}Hg				+2
	46	60		
	38	51		+2
	33	42		-3
$^{251}_{98}\text{Cf}$				